

KALINA, O.S. (Ryazan')

Pathological anatomy of congenital toxoplasmosis. Arkh. pat. 27
no.5:48-52 '65. (MIRA 18:5)

1. Kafedra patologicheskoy anatomii (zav. - prof. V.K.Belatskiy)
Ryazanskogo meditsinskogo instituta imeni Pavlova, kafedra
infektsionnykh bolezney (zav. - prof. L.K.Korovitskiy) Odesskogo
meditsinskogo instituta imeni Pirogova i 1-ya gorodskaya Odeselskaya
infektsionnaya bol'ница (glavnnyy vrach L.T.Zhidovlenko).

"APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5

KALINA, P., mayor.

Field radio switchboard for 14 directions. Voen. sviaz. 16 no.5;
41-44 My '58. (MIRA 11:5)

(Radio, Shortwave)

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5"

124-57-2-2569

Translation from: Referativnyy zhurnal, Mekhanika, 1957, Nr 2, p 148 (USSR)

AUTHORS: Korsakevich, N. I., Ofengenden, R. G., Kalina, P. N.

TITLE: Measuring Equipment for the Static Recording of the Stressed State of Articles (Izmeritel'nyye ustroystva dlya staticheskoy registratsii napryazhennosti detaley)

PERIODICAL: Nauch. tr. In-ta mashinoved. i s.-kh. mekhan. AN UkrSSR.
1955, Vol 5, pp 51-61

ABSTRACT: The analysis of the results of an experimental determination of the stressed state of any machine part over a sufficient prolonged period of time concludes in the sorting out of a large number of measured quantities according to their magnitudes and in the determination of the statistical distribution of the quantities that characterize the operating conditions of the part. The paper describes the operating principle of an electronic device for the automatic determination of the extremal values of the measured quantities. The input consists of an electric voltage which characterizes the measured parameter. The device automatically segregates the input voltages into six sub-ranges, which are equipped to transmit a signal to the

Card 1/2

124-57-2-2569

Measuring Equipment for the Static Recording of the Stressed State (cont.)

computing link, if an extremal value of the quantity will occur in the given sub-range. The computers of each sub-range count the number of values and upon completion of a test immediately provide an account of the number of the extremal values contained in the given sub-range. An example is adduced, showing the analysis of a generic curve and the determination of the maximal and minimal values thereof. The equipment described includes electromechanical computers capable of utilizing impulses lasting longer than 1/25 sec. It is possible, however, that computers be used which are capable of utilizing impulses lasting 1/200 sec and even less. A brief description is given of equipment having an analogous purpose, developed at the Institut stroitel'noy mekhaniki AN UkrSSR (Institute of Structural Mechanics, Academy of Sciences, Ukrainian SSR).

1. Recording devices--Performance 2. Stress analysis N. P. Rayevskiy

Card 2/2

KALINA, R.

"Specialization in the animal production as a means to achieve a higher labor productivity." P. 124

VESTNIK. Praha, Czechoslovakia; Vol. 6, No.3, 1959

Monthly List of East European Accession Index (EEAI), Library of Congress, Vol. 8, No. 7, July, 1959

Unclassified

KALINA, T.

Algology in the Czechoslovak Socialist Republic. Ukr. bot. zhur.
20 no.4:114-117 '63. (MIRA 17:4)

1. Charles University, Prague.

KHROMOV, B.M. (Leningrad); KOZLOVA, A.V.; KALINA, V.I.; ZADGENIDZE, G.A.;
FILIPPOVA, V.A.

Book reviews. Med. rad. 10 no.11:84-91 N '65.

(MIRA 19:1)

KALINA, V. K.

Cand Med Sci - (diss) "Effect of general roentgen radiation on biocolloids and on the functional activity of the thyroid gland." Khar'kov, 1961. 13 pp; (Ministry of Public Health Ukrainian SSR, Khar'kov State Med Inst); 200 copies; price not given; (KL, 7-61 sup, 259)

KALINA, V.K.

Change in the secretory activity of the thyroid gland following
whole-body X-ray irradiation. Radiobiologija 5 no.4:525-527 '65.

1. Kiyevskiy institut usovershenstvovaniya vrachey.
(MIRA 18:9)

KALINA, V.

"A Review of G. Gunning's article 'Tonsilectomy and Poliomyelitis'", Vest. Oto-rino-laringol., No. 1, 1948.

KALINA, V.

"Review of E. I. Pulaski's and Ch. S. Mathews' article 'Uses of Streptomycin,'" Vest. Oto-rino-laringol, No 1, 1949.

KALINA, V.

"Review of E. Fowler's and E. Seligman's Article 'Streptomycin,'" Vest. Oto-rino-laringol. No 1, 1949.

KALINA, V.K.

Volumetric automatic dispenser for radioactive liquids. Med. rad.
9 no.6:51-54 Je '64. (MIRA 18:2)

1. Kafedra meditsinskoy radiologii (zav.- prof. N.F. Lipkan)
Kiyevskogo instituta usovershenstvovaniya vrachey (rektor - dotsent
M.N. Umovist).

KALINA, V. O.

33591. Materialy K Voprosu O Prioritete I Rolt Otechestvennykh Otolaringologov V Razrabotke Khirurgicheskikh Metodov Lecheniya Zabolevaniy Ukrha, Nosa I Gorla. Vestnik Otorinolaringologii, 1949, No. 5, c. 3-16. Bibliogr: (Ukazatel' Literatury Po Khirurgii Nosa I Gorla). (Ukazatel' Literatury Po Zhirurgii Ukrha), c. 13-16

SO: Letopis'nykh Statey, Vol. 45, Moskva, 1949

KALINA, V. O.

Contemporary treatment of otogenic cerebral abscesses. Vest.
otorinolaryngologicheskaya. 12:4, July-Aug. 50. p. 76-83

1. Moscow.

CLML 19, 5, Nov., 1950

PLATONOV~~A~~, Ye.P.; KALINA, V.O.

Headache in acute, chronic paranasal sinusitis. Uchen. zapiski vtor.
moskov. med. Inst. Stalina Vol 2:114-119 1951. (CML 21:4)

1. Candidate Medical Sciences for Platonova. 2. Department of Nervous
Diseases (Director--Prof. A.M. Grinshteyn, Active Member of the Academy
of Medical Sciences USSR) and Clinic for Diseases of the Ear, Nose, and
Throat (Director--Prof. B.S. Preobrazhenskiy).

KALINA, V. O.

USSR/Medicine Encephalitis

Nov/Dec 53

"Nonpurulent Encephalitis of Aural Origin," V. O. Kalina, Cand Med Sci, Clinic of Nose, Throat, and Ear Disease, 2nd Moscow Med Inst im I. V. Stalin

Vest Otorinolaring, No 6, pp 35-40

Describes clinical observations of otogenic encephalitis during 1940-1950. Ascribes the prevalence of non-purulent conditions to the use of antibiotics and sulfa drugs in the treatment of otitis media and related conditions. Asserts that otogenic encephalitis is a mere stage of an inflammatory and infectious process in the

272T25

brain tissues, which, under certain unfavorable conditions, may become purulent. Names the condition of the central nervous system as an important factor in the development of purulent conditions.

KALINA, V.O., kandidat meditsinskikh nauk; PREOBRAZHENSKIY, B.S., professor, doyent,
viteznyy chlen Akademii meditsinskikh nauk SSSR, direktor.

Otogenous sinus thrombosis with symptoms of cerebral abscess. Vest. oto-rin. 15
no. 4:29-33 Jl-Ag '53. (MIRA 6:9)

1. Klinika bolezney ucha, gorla i nosa II Moskovskogo meditsinskogo instituta
im. I.V.Stalina. 2. Akademiya meditsinskikh nauk SSSR (for Preobrazhenskiy).
(Thrombosis) (Brain--Abscess)

KALINA, V.O., kandidat meditsinskikh nauk

Otogenous meningitis with symptoms of cerebral abscess. Vest.
oto-rin. 16 no.2:53-57 Mr-Ap '54. (MLRA 7:6)

1. Iz kliniki bolesney ukha, gorla i nosa (dir. chlen Akademii
meditsinskikh nauk SSSR prof. V.S.Preobrazhenskiy) II Moskov-
skogo meditsinskogo instituta imeni I.V.Stalina.

(MENINGITIS, differential diagnosis,
*brain abscess, otogenous meningitis)

(BRAIN, abscess,

*differ. diag. from otogenous meningitis)

(ABSCESS,

*brain, differ. diag. from otogenous meningitis)

KALIN, V. O.

Name: KALIN, Valentin Osipovich

Dissertation: Peculiarities of the Pathogenesis and
clinic of otogenous brain abscesses,
nonpurulent encephalitis, and
arachnoiditis

Degree: Doc Med Sci

Affiliation: Not indicated

Defense Date, Place: 15 Jun 56, Council of Dept of Clinical
Medicine Acad Sci USSR

Certification Date: 16 Mar 57

Source: BMVO 13/57

KALINA, Valentin Osipovich, red.; POGOSOV, V.S., red.

[Problems in theoretical and clinical otiatrics] Voprosy teoreticheskoi i klinicheskoi otiatrii. Moskva, 1960. 190 p.
(MIRA 13:12)

i. Moscow. TSentral'nyy institut usovershenstvovaniya vrachey.
(EAR--DISEASES)

KALINA, V. O., doktor med. nauk; SKURKOVICH, G. V.

Some problems in diagnosing cancerous tumors of the esophagus.
Vest. otorin. no.2:59-62 '62. (MIRA 15:2)

1. Iz Otorinolaringologicheskoy kafedry (zav. - prof. I. I. Potapov) TSentral'nogo instituta usovershenstvovaniya vrachey i iz Gorodskoy klinicheskoy infektsionnoy bol'nitsy No. 2, Moskva.

(ESOPHAGUS--CANCER)

POTAPOV, Ivan Ivanovich, prof.; ZBEROVSKAYA, Nina Viktorovna;
KALINA, Valentin Osipovich; VOLKOV, Yu.N., red.; PARAKHINA,
N.L., tekhn. red.

[Tympanoplasty] Timpanoplastika. Moskva, Medgiz, 1963. 166 p.
(MIRA 16:12)
(TYMPANAL ORGAN—SURGERY)

FEL'DMAN, Aleksandr Isidorovich [deceased]; VUL'FSO, Solomon Isaakovich; KALINA, V.O., red.

[Diseases of the ear and upper respiratory tract in children] Bolezni ucha i verkhnikh dykhatel'nykh putei v detskom vozraste. 6-3 izd. perer. i dop. Moskva, Meditsina, 1964. 399 p. (MIRA 17:5)

GLADKOV, Aleksandr Aleksandrovich; KALINA, V.O., red.

[Diseases of the nose, throat and ear] Bolezni nosa,
gorla i ukha. Moskva, Meditsina, 1965. 365 p.
(MIRA 18;2)

KALINA, V.O.; KACHOROVSKAYA, I.B.; MERKOVA, M.A.; GINZBURG, M.B.

Sequelae of radiotherapy of cancer of the larynx. Med. rad.
9 no.11:3-7 N '64. (MIRA 18:9)

1. Nauchno-issledovatel'skiy rentgeno-radiologicheskiy institut
Ministerstva zdravookhraneniya RSFSR.

KALINA, Yu.P.; BABYRIN, V.V.

Determination of indium in process solutions. Zav. lab. 31 no.8:
946 '65. (MIRA 18:9)

MACHALSKI, Marek; KALINA, Zbigniew; FOREMNY, Zbigniew; PATER, Janusz

Postbulbar duodenal spasm syndrome. Pol. tyg. lek. 20 no.38:
1422-1424 20 S '65.

1. Z I Kliniki Chorob Wewnetrznych Slaskiej AM w Katowicach
(Kierownik: prof. dr. Jozef Japa) i z Zakladu Radiologii
P.M.R.N. w Katowicach (Kierownik: dr. Maria Mromlinska).

POLAND

GALA, Jerzy and KALINA, Zbigniew, First Clinic of Internal Diseases (I Klinika Chorob Wewnętrznych), Sl. AM [Slaska Akademia Medyczna, Silesian Medical Academy] in Katowice (Director: Prof. Dr. J. JAPA)

"Acute Necrosis of the Small Intestine During Infectious Polyneuroradiculitis Treated with Encortone."

Warsaw-Krakow, Przeglad Lekarski, Vol 19, Ser II, No 3, [24 Mar] 63, pp 188-189.

Abstract: [Authors' English summary] The authors present a case of the Guillain-Barre-Strohl syndrome in a 53-year old male, in whom death ensued following administration of encortone, as a result of perforating ulceration of the small intestine and consequent diffuse peritonitis. Of the 13 references, one is French, two are English, and ten are in Polish.

1/1

KALINA-ZHIKHAREVA, V.I.

137-58-5-11192

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 5, p 327 (USSR)

AUTHORS: Gurevich, A. B., Kalina-Zikhareva, V. I.

TITLE: Employment of Cationites and of Trilonometric Titration for Determination of Arsenic in High-arsenic Alloys (Opredeleniye mysh'yaka v vysokomysh'yakovistykh splavakh s primeneniem kationitov i trilonometricheskogo titrovaniya)

PERIODICAL: Tr. Nauchno-tekhn. o-va chernoy metallurgii. Ukr. resp. pravl., 1956, Vol 4, pp 127-130. Comments, pp 131-137

ABSTRACT: As is separated from Fe by means of passing a solution of the alloy through an ion exchanger containing 60 g of sulfocarbon or 40 g of KU-2. The As is precipitated in the solution by a magnesia-ammonia mixture. $MgNH_4AsO_4$ is filtered out and is dissolved in HCl (1:1); after adding NH_4OH , a buffer solution, and an indicator (acidic, dark-blue Cr), the As is titrated with a solution of trilon B. Another method of titrating As with trilon B is also described. The results of determination of As in ferroarsenic are presented in a tabulated form.

Card 1/1 1. Arsenic compounds 2. Arsenic--Determination P. K.
 3. Titration--Applications 4. Ions--Applications

KALINA-ZHIPHAREVA - V. I.

KALINA, V.I.

137-58-5-11152

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 5, p 322 (USSR)

AUTHORS: Kalina, V.I., Leve, N.F.

TITLE: Employment of Cationites and Trilon B in the Analysis of Open-hearth Slags with a High Content of Phosphorus Pentoxide
(Analiz martenovskogo shlaka s vysokim soderzhaniyem pyatikisi fosfora s primeneniem kationitov i trilona B)

PERIODICAL: Byul. nauchno-tekh. inform. Ukr. n.-i. in-t metallov,
1957, Nr 3, pp 92-94

ABSTRACT: In analyzing P-slags, the separation of the PO_4^{3-} anion from the cations of the second and third analytic groups is accomplished by a chromatographic method, which utilizes the cationites "vofatite"-P and sulfocarbon. After dissolving 1 g of slag in HCl, several drops of HNO_3 are added, and siliceous acid is separated in the usual manner. The solution is transferred to a graduated cylinder and an aliquot portion, diluted with H_2O , is passed through a column containing "vofatite"-P. All cations are absorbed by the vofatite, while the P remains in the filtrate, where it is determined by the volumetric-alkalimetric method. The cations are washed out of the vofatite-P with a solution of

Card 1/2

137-58-5-11152

Employment of Cationites (cont.)

HCl (1:4), and are then analyzed (for Fe, Mn, Al, Mg, and Ca). Mg is determined by the complexometric method with trilon B and a dark-blue Cr indicator. The analysis employing sulfocarbon is analogous to the procedure described, but requires a more thorough purification of the sulfocarbon.

A. M.

1. Slags--Analysis
2. Ions--Applications
3. Phosphorus pentoxide--Determination
4. Magnesium--Determination

Card 2/2

MURADKHANYAN, L.K., kand. sel'skokhoz. nauk; DROZDOV, V.N.; KOVALEV, A.T.;
KALINCHENKO, V.I.

Machines and attachments for the placement of mineral fertilizers.
Zemledelie 27 no.4:32-36 Ap '65. (MIRA 18:4)

1. Nauchno-issledovatel'skiy institut sel'skogo khozyaystva
tsentral'nykh rayonov nechernozemnoy polosy.

Kalinashenko, V. P.

55 R

Investigation of the reactivity of aromatic compounds by the method of isotopic exchange with deuterium oxide. V. P. Kalinashenko, Yu. M. Vinogradov, and A. V. Tsvetkov. Zvezdnyi Akad. Nauk S.S.R. 91, 171-180 (1959). J. C.A. 49, 9305b. Data on H-D exchange between aromatic compounds and liquid KBr solution are summarized. Such experiments show differences in the velocity of nonreactive H atoms of the aromatic compounds more sharply than do experiments with NO_2 (J. C.A. 48, 10412). These differences might be used for separation of compounds partially differentiated in the aromatic nuclei. Selected data follow: given in the parentheses reagent, temp., time, μ , and k (hr. of 10% of D atoms exchanged for H, $k = \frac{1}{2} \ln 2 / t$) unless rate constant or time is given.

- C₆H₆, room temp., 160-175 hr., 0.2-0.4, 1.2×10^{-4} , biphenyl, 25°, 30 min., 4.5×10^{-4} for ortho and 7.5×10^{-4} for meta; nitrobenzene, 25°, 26 min., 7.5×10^{-4} ; nitroanisole, 25°, 20 min., 10.5×10^{-4} ; acetophenone, 0°, 30 min., 10, & not given; biphenyl, 25°, 1 hr., 1.2 (in 2-14 days reaction constant value); 2×10^{-4} for the para-H and 2×10^{-3} for ortho-H; p-terphenyl, 25°, 1 hr., 2-14 days reaction constant value; 7×10^{-4} for para-H and 6×10^{-4} for ortho-H; DPH (R = Me, Et, Me_2CH , Me_2N , or Ph), 26°, 2 hrs., 1-1.5 (values for equilibrium given in the paper); 3×10^{-4} for para-H and 3×10^{-3} for ortho-H and 2.5×10^{-2} for methyl- O_2-NO_2 ; diethyl ether, 25°, 1 hr., 0.2 (in 2.2 hrs., 12.5 in 24 hr.), 6×10^{-4} for para-H; triphenylmethane, 25°, 1 hr., 0.05 (10.0 in 454 hrs.); 8×10^{-4} for para-H; 1,2-dimethylcyclohexane, 25°, 1 hr., 0.7, & not given; 1,4-dimethylcyclohexane, 25°, 1 hr., 1.7, & not given; Indene, pyridine, 25°, 1 hr., 0.05 (in 24 hr.); 4-nitroanisole in 1 hr.

KALINCHIKO, V. R.

"Investigating Isotopic Hydrogen Exchange in Aromatic Compounds in a Medium of Liquid Deuterium Bromide." Cand Chem Sci, Sci Res Physicochemical Inst imeni L. Ya. Karpov, Min Chemical Industry, Moscow, 1955. (KL, No 18, Apr 55)

SO: Sum. No. 704, 2 Nov 55 - Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (16).

KALINACHENKO, V. R.

USSR/Chemistry - Physical chemistry

Card 1/1 Pub. 22 - 33/53

Authors : Shatenshteyn, A. I.; Zhdanova, K. I.; Vinogradov, I. N.; and Kalinachenko, V. R.

Title : About oxygen catalysis of isotopic hydrogen exchange in a liquid deuterium bromide medium

Periodical : Dok. AN SSSR 102/4, 779-782, Jun 1, 1955

Abstract : Experiments were carried out with twenty substances containing halide, nitro, cyan, sulfo or carboxyl groups to determine the solubility of these substances in liquid DBr in the presence of AlBr₃ and their stability and behavior in the solution. It was found that the isotopic hydrogen exchange observed in certain saturated hydrocarbons mixed with D₂SO₄, DBr - AlBr₃, DF and DF - BF₃ is evidently connected with the exclusively high acidity of the medium. The results of the experiments carried out at room temperature are tabulated. Twenty-two references: 12 USA, 9 USSR and 1 German (1935-1954). Table.

Institution : The L. Ya. Karpov Sc. Res. Physicochemical Inst.

Presented by : Academician V. A. Kargin, February 24, 1955

KALINACHENKO, V.

USSR/Organic Chemistry. Theoretical and General
Questions of Organic Chemistry.

E-1

Abs Jour : Ref Zhur - Khimiya, No. 8, 1957, 26613.

Author : Kololev, A.; Shatenshteyn, A., Yurygina,
Ye., Kalinachenko, V., Alikhanov, P.

Inst Title : Isomerization of Monodeuteronaphthalenes.

Orig Pub : Zh. obshch. khimii, 1956, 26, No. 6, 1666 -
1672.

Abstract : The question of the possibility of transposition of α -H and β -H in the naphthalene molecule was investigated by the method of deuterium interchange. It is shown that if vapors of α -deuteronaphthalene, as well of β -deuteronaphthalene (I and II) in a flow of nitrogen are passing above silica gel at 420°,

Card 1/3

KALINACHENKO, V.R.

SHATENSHTEYN, A.I.; KALINACHENKO, V.R.; VARSHAVSKIY, Ya.M.

Hydrogen exchange between benzene and naphthalene derivatives and liquid deuterium bromide [with English summary in insert]. Zhur. fiz.khim. 30 no.9:2098-2105 § '56. (MIRA 9:12)

1. Fiziko-khimicheskiy institut imeni L.Ya. Karpova, Moskva.
(Hydrocarbons) (Hydrobromic acid-d)

KALINACHENKO, V. R.

76-1-22/32

AUTHORS: Shatenshteyn, A. I. , Peregudov, G. V. , Izrailevich, Ye. A. ,
Kalinachenko, V. R.

TITLE: Preparation of Some Deuterated Aromatic Hydrocarbons and Their
Raman Spectra (Polucheniye nekotorykh deuterirovannykh aromaticheskikh
uglevodorodov i ikh spektry kombinatsionnogo rasseyaniya)

PERIODICAL: Zhurnal Fizicheskoy Khimii, 1958, Vol. 32, Nr 1, pp.146-151 (USSR)

ABSTRACT: Some known preparations, as well as some ones of deuterated aromatic hydrocarbons still not being mentioned in literature were obtained, and their Raman spectra were measured by means of isotope exchange of hydrogen with $ND_3 + KND_2$ or with liquid DBr. A comparison between the Raman spectra of hexadeuterobenzene- and octadeuteronaphthalene preparations and those from the references 11, 12, 14 and 15 proves the identity of all preparations and fully confirms the approbation of the new methods of preparation of deuterated hydrocarbons. The advantages of these new methods in relation to those of other authors are enumerated: rapid reaction, the solvent is easily to be removed, possibility of a complete deuteration of various aromatic-, aliphatic-aromatic- and ethylene-hydrocarbons, as well as many other organic compounds. The advantage on occasion of the deuteration by means of liquid DBr in relation

Card 1/4

76-1-22/32

Preparation of Some Deuterated Aromatic Hydrocarbons and Their Raman Spectra

to that one by means of liquid ND₃ is the higher coefficient of the deuteration distribution between the Cu- and NBr bonds ($\alpha = 3,0$) in comparison to $\alpha = 0,9$ in CH- and NH bonds (see reference 18 and 21). In the presence of an equal quantity of heavy water, 25 times more of DBr than of ND₃ is obtained, besides. The pure benzene- and toluene preparations placed at the disposal by A. L. Libernab served as initial substances. The liquid ND₃ was prepared by the action of D₂O (99,6 atm.% D) upon Mg₃N₂ (reference 2), whilst the liquid DBr was produced synthetically from the elements (reference 24). The technique of the experiment has been described in these references. Presently, the representation of the deuterio-ammonia is simplified: Mg₃N₂ and an ampule with heavy water are put into a steel balloon. A valve is screwed in into the latter one. By means of destroying the ampule the reaction between Mg₃N₂ and D₂O is liberated. On occasion of the hydrogen exchange the substances exist in the solution. As a rule the potassium amide concentration is not great (0,02 N). The exchange reaction was carried out once more with new solvent portions at room temperature during a period, which guarantees the obtaining of the equilibrium in the exchange reaction. On occasion of the experiments with C₁₀H₈ the number of the ND₃ mols amounted to 50-150 per mol of substance, whilst on occasion of the experiment with benzene

Card 2/4

76-1-22/32

Preparation of Some Deuterated Aromatic Hydrocarbons and Their Raman Spectra

and toluene per mol of substance only 20-40 mol of the solvent was taken, because these hydrocarbons (especially benzene) on occasion of distilling after the experiment are easily entrained by ammonia. In order to obtain preparations by means of isotope exchange with liquid DBr 1 g of the substance was dissolved in 15 - 22 g of liquid DBr. After evaporation of the solvent the liquid substances were distilled (above CuSO₄, in order to remove the ND₃ traces, or above Na, in order to bind DBr). Solid substances were distilled 2 - 3 times.

The spectra of the combined scattering (Raman spectra) were measured by means of a two-prism-spectrograph "Huet" with relative aperture 1 : 4,7 and a dispersion of 100 cm⁻¹ in the range of 4358 Å. In the following work the computations, and the interpretation of the spectra are given, and the values of the frequencies are precisely. The Raman spectra of following substances were measured: C₆D₆, C₁₀D₈, C₁₂D₁₀, C₆D₅CD₃, 1,4,5,8-C₁₀D₄H₄, 2,3,6,7-C₁₀D₄H₄, 2,4,6,2',4',6'-C₁₂D₆H₄ and 3,5,3',5'-C₁₂D₄H₆.

The authors were advised by G. S. Landsberg, Academician, and Professor P. A. Bazhulin. P. N. Manochkina assisted. The density of

Card 3/4

76-1-22/32

Preparation of Some Deuterated Hydrocarbons and Their Raman Spectra

the preparations was measured by Yu. I. Antonchik. The preparations were placed at disposal by A. L. Liberman. There are 2 tables, and 24 references, 9 of which are Slavic.

ASSOCIATION: Physical-Chemical Institute imeni L. Ya. Karpov. AS USSR. Institute for Physics imeni P. N. Lebedev, Moscow
(Fiziko-khimicheskiy institut im. L. Ya. Karpova. Akademiya nauk SSSR. Fizicheskiy institut im. P. N. Lebedeva. Moskva)

SUBMITTED: October 31, 1956

AVAILABLE: Library of Congress

Card 4/4

'5 (2,3)

SOV/79-29-3-21/61

AUTHORS: Shatenshteyn, A. I., Kalinachenko, V. R., Yurygina, Ye., N., Basanova, V. M.

TITLE: Deuteron Exchange Between Liquid DBr and Phenylated Alkanes
(Deyterroobmen mezhdu zhidkimi DBr i fenilirovannymi alkanami)

PERIODICAL: Zhurnal obshchey khimii, 1959, Vol 29, Nr 3, pp 849-855 (USSR)

ABSTRACT: The reaction rate of the electrophilic substitution of hydrogen in alkyl benzenes [(of the chlorination (Refs 1,2), bromination (Ref 3), nitration (Ref 4) and alkylation according to Friedel-Crafts (Ref 5)) decreases in the following order:
 $C_6H_5CH_3 > C_6H_5C_2H_5 > C_6H_5CH(CH_3)_2 > C_6H_5C(CH_3)_3$. This is explained (Refs 6,7) by the effect of "superconjugation" ($\sigma\pi$ -conjugation). It may be assumed in an analogous way that the rate of the corresponding reactions, under participation of the polyphenylated alkanes, also depends on the ratio of the number of the α -CH-bonds to the number of the aromatic rings. If it is, however, taken into account that the bromination rate of the alkyl benzenes depends on the ramification of the carbon chain not only on the α -, but also on the β -carbon atom of the alkyl

Card 1/3

SOV/79-29-3-21/61

Deuteron Exchange Between Liquid DBr and Phenylated Alkanes

group (Ref 3) it is not impossible that in the reactions of the electrophilic substitution of hydrogen in other phenylated alkanes the ratio between the number of rings and the number of the more remote CH-bonds is of importance. In order to prove the correctness of these assumptions the authors investigated the deuteron exchange between the polyphenylated alkanes and liquid DBr (Refs 8,9). Its mechanism is closely related with the mechanism of the chemical reactions of the electrophilic substitution of hydrogen (Ref 10). Some results were already earlier published (Ref 11). Experiments of this kind were carried out with the following hydrocarbons: diphenyl, triphenyl, tetraphenyl methane, fluorene, dibenzyl, sym.-tetraphenyl ethane, 1,1,1-triphenyl ethane, 1,3-diphenyl propane, 1,4-diphenyl butane and 1,5-diphenyl pentane. Thus it was demonstrated that the phenyl rings separated by the carbon atom (in tetraphenyl methane) are of mutual influence as regards the reactivity. It is compared with the influence exerted by the effect of the $\pi\pi$ - and $\sigma\pi$ -conjugation upon the reactivity of the aromatic ring. There are 2 tables and 36 references, 16 of which are Soviet.

Card 2/3

SOV/79-29-3-21/61

Deuteron Exchange Between Liquid DBr and Phenylated Alkanes

ASSOCIATION: Fiziko-khimicheskiy institut imeni L. Ya. Karpova i Nauchno-issledovatel'skiy institut poluproduktov i krasiteley
(Physico-Chemical Institute imeni L. Ya. Karpov and Scientific Research Institute of Semiproducts and Dyes)

SUBMITTED: February 10, 1958

Card 3/3

KUZ'MIN, N.M.; BELYAYEV, V.P.; KALINACHENKO, V.R.; YAKIMENKO, L.M.

Chemical-spectral method of the analysis of high-purity
alkalies. Zav. lab. 29 no.6:691-692 '63. (MIRA 16:6)

(Alkalies) (Spectrochemistry)

KALINAK, Iosif, novator

Device for feeding mortars to vertical and horizontal joints of
masonry walls and seams of reinforced concrete wall elements.
Suggested by Iosif Kalinak. Rats.i izobr.predl.v stroi. no.13:
96-98 '59. (MIRA 13:6)

1. Po materialam Instituta tekhnicheskoy dokumentatsii Rumynskoy
Narodnoy Respubliki.
(Mortar)

YANKAUSKAS, M. M. [Jankauskas, M.]; KALINAUSKAS, A. T. [Kalinauskas, A.]

Maximum percentage of reinforced steel in bar-reinforced concrete beams. Liet ak darbai B no.1:243-255 '61. (EEAI 10:9)

1. Institut stroitel'stva i arkitektury Akademii nauk Litovskoy SSR.

(Reinforced concrete)

YANKAUSKAS, M.M. [Jankauskas, M.]; KALINAUSKAS, A.T.

Crack resistance analysis of prestressed reinforced concrete
elements subjected to bending. Trudy AN Lit. SSR. Ser.B.
no.1:235-242 '62. (MIRA 17:8)

1. Institut stroitel'stva i arkhitektury AN Litovskoy SSR.

YANKAUSKAS, M.M. [Jankauskas, M.]; KALINAUSKAS, A.T.

Experimental study of the effect of prestressing on the carrying capacity
of a reinforced concrete component having a rectangular cross section and
subjected to bending. Trudy AN Lit. SSR. Ser.B no.1:193-210 '65.
(MIRA 18:7)

1. Institut stroitel'stva i arkhitektury AN Litovskoy SSR.

KALININSKENS, B. I.

Sov/77-4-2-1578

21(9) 23 (5)

Lyalikov, K.S.

AUTHOR: Successes of Soviet Electrophotography (Uspishi sovetskoj elektrofotografii) A Technical Conference on Questions of Electrophotography (Nauchno-tekhnicheskaya konferentsiya po voprosam elektrofotografii)

TITLE: PERIODICAL: Zhurnal nauchnoj i prakticheskoy fotografii i kinematografii, 1959, Vol. 4, No. 2, pp 149-152 (USSR)

ABSTRACT: This is an account of a scientific and technical conference on electrophotography, the first to be held in the Soviet Union and abroad, in the world. It was organized in Vilnius on December 16-19, 1958 by the Soviet Standardization Commission of the Lithuanian SSR (Council for National Economy of the Lithuanian SSR), the Council for Standardization, Technical Committee of the Scientific Research Institute of Scientific and Technical Committee of the Lithuanian SSR (State Scientific and Technical Committee of the Ministry of Universities of Lithuania SSR) and of the Council of Higher Education of the Lithuanian SSR, and the Kaunas-14 Research Institute of Electrographical Scientific Research Institute of Electrophotography. The conference attended by over 300 scientific workers was opened by the Deputy Chairman of the Council for National Economy of the Lithuanian SSR P.A. Kalnietis, at which the director of the Institute of Electrophotography, V.I. Zhalevich, reviewed the state of electrophotography in the field of development of the field should be pursued. He stated that research in this field should be carried out along the following lines: a) a search for new photo-active materials with high dark resistance; b) physical research into the interlayer photoeffect; c) development of photoconductor layers; d) development of the theory of the electrophoto-optic process. K.S. Lyalikov (Speaking also for O.G. Ignorov) gave a report in which he succeeded determining the light sensitivity of electrophotographic layers in GDR units. N.N. Plavkin (Speaking also for I.I. Chilieva, L.I. Sunko, V.M. Markovich, E.I. Kalininskas and O.M. Sutkaitis) reported on some research on the sensitization of semiconductor in electrophotographic layers. V.Z. Zaitin gave a report on highly sensitive electrophotographic layers as an electronic photocopier device. B.R. Erdogdu gave a report on the formation processes of the latent electrophotographic image on the basis of the contact photographic image on the basis of the contact photographic image on the basis of the contact photographic image. He also described the design of an electrophotocopying device for determining sensitivity by the retention period of charge on the surface of the layer, and the circuit of an electrophotographic copying device. J. Jukuryte finished describing the latter and then spoke on the mechanics and kinetics of the development of the latent electrophotographic image in liquid developer.

Card 3/10

SU/77-4-215/8

Successes of Soviet Electrophotography: A Scientific and Technical Conference on Questions of Electrophotography

K.M. Vinogradov described some of the features of the cascade and liquid methods of electrophotographic development. Yu.Ye. Tarasevich derived his report to the criterion of light sensitivity of the electrophotographic process. After the reports, a discussion took place on methods of determining the light sensitivity of electrophotographic layers. A.V. Cherepanov spoke on the prospects of developing polymeric processes dealing with electric and magnetic forces. O.I. Grozov (speaking also for I.I. Zhilevich, A.A. Sushko, V.A. Gordeyev, A.S. Pausha and Yu. I. Kavaliaylo) reported the development of electrophotographic reproducing equipment. A.G. Pausha (speaking also for I.I. Zhilevich, A.S. Botisovich, M.H. Gal'rikov and N.I. Rukhankova) reported on the use of electrophotographic methods in recording oscillographs and other recording instruments.

V.P. Chishikyan (speaking also for L.M. Sulian) spoke on the possibility of automatically recording images from electron-beam tubes. G.S. Kozlov (speaking also for N.N. Markovich, V.I. Chikishev, B.I. Kalinushkin, N.E. Mayakovskii, V. I. Chikishev and K.A. Monirul'sh) gave a detailed description of laboratory and machine methods of producing photoacoustic papers (fine orides were used). A.A. Smirnov (speaking also for I.I. Zhilevich, O.V. Grozov, V.A. Gordeyev, N.V. Fedotov and F.M. Guf) described a laboratory and industrial machine for producing photosemiconductor paper. F.V. Chishikyan (speaking also for Ya.A. Osman) reported on a method of examining electrophotographic materials using an A/C bridge. S.I. Khotzhanovich (speaking also for I.I. Gilens and I.G. Shchelkovich) spoke on developing materials for electrophotography and letter recognition, including development methods, a reverse image. Khotzhanovich provided methods of measuring the electropotential of electrophotographic layers. He stressed that the dielectric voltage across the layer should not be placed above the dielectric potential at this causes short-circuiting. I.Y. Kul'ovskii (speaking also for A.G. Gordeyev, A.A. Orl'gov and S. Chernets) spoke on the practice of producing television papers in an electrostatic field, and showed samples produced by the Gribachinsk paper factory.

Yu. M. Sosulin then gave a historical review of the development of electrophotographic methods in which he paid tribute to the work of the Scientific Research Institute of Electrophotography in Tbilisi and the Institut Poligraficheskogo Mashinostroyeniya (Polygraphic Machine-building Institute) (Tbilisi). Debates were then held

Card 6/10

on methods of measuring the potential of charged electro-photographic layers. The vibration pick-up most-used was shown in B.I. Tikhonov's report to be not always accurate. S.G. Graniashin stated that the bad influence of the oscillating electrode can be eliminated if the electrode probe above its surface is fixed and the pick-up is connected to it by a shielded cable. In the debate on V.-L. Masirovsky's report, it was stated that the research of Academician N. Terenin and Yu.K. Papulin reviewed some of the results of the use of electro-photographic methods in radiography. I.-I. Ruzhko (speaking also for I.I. Zhilevich, I.Z. Pavlin, Yu.K. Tishchenko and Yu.A. Zibutov) reported on relaxation processes in semiconductor layers, using a vibration electrode. In F. V. Shchukin's paper, he gave a report on research on some physical properties of the polycrystalline layers of anodous titanium. M.I. Klyklyavichus spoke on some of the photoelectric properties of the film (Soviet 225, the absorption maximum of the latter is about 900 nm).

S.M. Ilyushin reported on methods of obtaining selenium and thermosensitive layers, including sublimation and thermal treatment; it was also found that the sensitivity of the layers increased after storage for 1.5 to 2 months at room temperature. P.M. Podliskalik (speaking also for S.G. Graniashin) spoke on research into the electrical properties of electrophotographic layers of amorphous selenium and powdered zinc oxide. N.K. Shil'nikov (speaking also for A.S. Naumovskiy) discussed the production of selenium layers and some of their properties. Finally, the following reports on ferrography were delivered: 1) B.Ia. Kostomarov, V.L. Zhdanov, Electrodeposition of Magnet-hard Alloys with GdMn Magnetic Characteristics; 2) I.Y. Butukov, Visualisation of Electromagnetic Oscillations by the Ferrographic Method; 3) V.G. Parshunov, Ferrography Recording of Fasible Images; 4) T.I. Zhilevich, I.I. Glik, B. Ye. Bubnov, I.I. Bernshteyn, A.F. Likhachev, Mock Experiments in Non-Pressure Ferromagnetic Stirring. There was also an exhibition showing the work of the Electrographic Institute.

The most important conclusion of the conference was that a solid approach had been made to the possibility of wide technical use of the methods of electrography. It was considered that although work in this field probably started only in 1955-56, it has made great strides in 10 years. It will certainly be easier to reproduce results already achieved than to be disappointed. According to Shchukin, one of the main reasons that the Americans took good care that no important information appeared in the literature available.

Card 10/10

KALINCHENKO, A.

KALINCHENKO, A.

Electric Coils

Elimination of breaks in coils. Radio 29 No. 3, 1952

Monthly List of Russian Accessions, Library of Congress, June 1952. UNCLASSIFIED.

AUTHORS: Kalinchenko, L.P., Strakhov, N.P.,
Kalinichenko, I.I. 32-1-7/55

TITLE: New Color Reaction for the Ascertainment and Determination of Beryllium With Chrome-Blue K (Novaya tsvetnaya reaktsiya dlya otkrytiya i opredeleniya berilliya s khromatsinim K).

PERIODICAL: Zavodskaya Laboratoriya, 1958, Vol. 24, Nr 1, pp. 22-23 (USSR)

ABSTRACT: Beryllium ions form a well soluble compound of bright-blue color with the acid chrome-blue K in the ammonia medium at pH=9-10. The effect produced by other ions (as e.g. Cu, Ni, Al, Co, Ba, Mg, Ca, Cd, Zn and others), which might produce colors with K, is eliminated by their addition to trilon. In this way it is possible to ascertain and quantitatively to determine beryllium. For the determination of beryllium a drop of the solution to be investigated is dropped onto filter paper; this is followed by a drop of ammonia buffer solution (20%), a drop of 0,1-n trilon B, and a drop of a 25% aqueous solution of acid chrome-blue K. If, after drying, a small blue or sky-blue spot forms in the center of the pink or violet-red spot, this indicates the presence of beryllium in the solution investigated. Otherwise, the same reaction may be observed in the test tube. If the beryllium content is not less than 10^{-7} g/ml, a pinkish

Card 1/2

New Color Reaction for the Ascertainment and
Determination of Beryllium With Chrome-Blue K

32-1-7/55

sky-blue color is obtained, and with a beryllium content of from $3 \cdot 10^{-8}$ to 10^{-7} g/ml the color will be bluish-violet. Within the range of $2 \cdot 10^{-6}$ g/ml this color is in accordance with the Lambert-Beer law. The maximum of light absorption in a pure reagent amounts to 580-590 μm and with the beryllium complex - 600-610 μm . Results are given in a table; a second table deals with ascertaining the presence of beryllium in bronze. There are 2 tables.

ASSOCIATION: Sverdlovsk Medical Institute and Ural Polytechnical Institute
im. S. M. Kirov (Sverdlovskiy meditsinskiy institut i
Ural'skiy politekhnicheskiy institut im. S.M.Kirova).

AVAILABLE: Library of Congress

Card 2/2 1. Beryllium-Determination

KALINCHENKO, L.P.; KALINCHENKO, I.I.

Complexometric determination of iryllium in copper alloys.

Trudy Ural.politekh.inat. no.130: '0-73 '63.

(MIRA 17:10)

KALINCHENKO, L.P.; KALINICHENKO, I.I.

Titrimetric determination of beryllium by means of sulfosalicylic acid. Zhur.anal.khim. 17 no.7:840-843 O '62. (MIRA 15:12)

1. Sverdlovsk Medical Institute and S.M.Kirov Ural Polytechnical Institute, Sverdlovsk.
(Beryllium—Analysis) (Salicylic acid)

Kalinchev, E.L.

82109
S/184/68/00/02/01/006

15. D110

AUTHORS: Kalinchev, V.E.L., Engineer, Rips, S.M., Candidate of Technical Sciences

TITLE: The Determination of the Cooling Time of Parts in the Mold in Injection-Molding of Plastics ✓

PERIODICAL: Khimicheskoye mashinostroyeniye, 1960, No 2, pp 22 - 26

TEXT: The article deals with the dynamics of the cooling process of parts in the mold, which has a direct effect on the efficiency of an injection molding machine. The determination of the full time of the injection cycle is complicated by the unsteady character of thermal conditions during the cooling process. The industrial experience accumulated and the authors' experiments show that the temperature of the mold does not change much during casting. The temperature of the mold changes within 1.5 - 5°C at a temperature of the injected mass within 295 - 200°C. An ample cooling water supply results in a drop of the mold temperature. The existing injection molding machines complete the injection within a short time (4.5 sec for pieces of 12 kg and 1.5 - 3 sec for lighter pieces) which is too short to record more or less sub-

Card 1/3

B2109
S/184/60/000/02/01/006

The Determination of the Cooling Time of Parts in the Mold in Injection-Molding of Plastics

stantial temperature changes in the center of the injected mass. For this reason it is assumed that the temperature in the center is constant and equal to the temperature of the injected mass. Experiments were carried out to determine the rate of cooling (M.A. Bochtarev and O.O. Baglayevskaya participated). A long polystyrene spiral, 20 mm wide and 5 mm thick, was used (coefficient of temperature conductivity $a = 2.5 \cdot 10^{-4} \text{ m}^2/\text{hour}$). One Chromel-Copel thermocouple was fixed in the center of the mass and another one was located 1 mm closer to the surface. The changes of temperature in time were recorded by an "OT24-51" (OT24-51) tensometric device. Temperatures of both mold sections were recorded by "MICP-01" (PSR4-01) electronic potentiometers. The temperature of the injected mass was determined in a nozzle of a special design, where the thermocouple could be fixed in the center of the mass to be injected (Figure 3). The experimental results agreed with theoretical calculations. The degree of divergence between calculation results obtained by the Fourier method and by the method of elementary heat balances and by the experimental investigation is shown by graphs (Figure 6) and Table 2. Besides temperature and pressure of the injected mass, the decisive factor in injection

Card 2/3

W

"APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5

SAGALAYEV, G.V.; KALINCHEV, E.L.

New molding machines. Plast.massy no.6:48-54 161. (MIRA 14:5)
(Plastics—Molding)

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5"

RABINOVICH, I.M.; KRICHEVER, I.S.; KALINCHEV, E.L.

New injection molding machine model TP-65. Plast.massy no.4:43-
45 '61. (MIRA 14:4)
(Plastics—Molding)

RABINOVICH, I.M.; KRICHEVER, I.S.; KALINCHEV, E.L.; MARAM, Ye.I.

New "TP-32" model molding machine. Plast.massy no.8:50-51 '62.
(MIRA 15:7)
(Plastics machinery)

KALINCHEV, E.L., inzh.; LEVIN, A.N., doktor tekhn.nauk, prof.

Calculating the force necessary for closing the molds of molding machines.
Khim.mashinostr. no.2:8-10 Mr-Ap '63. (MIRA 16:4)
(Plastics—Molding)

BEDILO, V.Ye.; KALINCHUK, I.G.; LISHBERGOV, V.D.; NIKOLAYEV, G.P.; TSOY, D.; SHCHUKINA, G.F. Prinimali uchastiye: KOLESNIKOV, V.P.; OSTAPENKO, P.V.; SEDOVA, M.P.; TKACHEV, M.V. DUGIN, Ye.V., otv.red.; RABINKOVA, L.K., red.izd-va; KOROVENKOVA, Z.A., tekhn.red.; SABITOV, A., tekhn.red.

[Types of mine cross section] Tipovye secheniya gornykh vyrabotok. Moskva, Gos.neuchno-tekhn.izd-vo lit-ry po gornomu delu. Vol.6. [Cross section of mines lined with steel arches and anchor bolting for 1-, 2- and 3-ton railroad cars] Secheniya vyrabotok, zakreplennikh stal'noi arochnoi i shtangovoi krep'iu, dlia 1-, 2- i 3-tonnykh vagonetok. 1960. 503 p. (MIRA 13:12)

1. Khar'kov. Gosudarstvennyy proyektnyy institut Yuzhgiproshakht. (Mine timbering)

PRIYEMCHENKO, A., polkovnik; KALINCHUK, M., podpolkovnik

Gadets learn to fight a battle. Voen.vest. 42 no.5:66-69
My '62. (MIRA 15:11)
(Attack and defense (Military science))

"APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5

KALINCHUK, P., podpolkovnik; SOLOV'YEV, A., podpolkovnik

Topography in the interests of tactics. Voen. vest 43 no.1:63-65 Ja
'64. (MIRA 17:1)

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5"

"APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5

KALINCHUK, P., podpolkovnik

Dissemination of technical information. Tyl i snab.Sov.Voor,Sil
21 no.1:30-32 Ja '61. (MIRA 14:6)
(Military education)

APPROVED FOR RELEASE: 03/20/2001

CIA-RDP86-00513R000620030005-5"

KALINENKO, I. G.

"The Controlled Change of Common Winter Wheat Into Rameose Wheat by Means of Training." Cand Biol Sci, All-Union Order of Labor Red Banner Selection and Genetics Inst imeni T. D. Lysenko, Odessa, 1955. (KL, No 12, Mar 55)

So: Sum. No 670, 29 Sept 55 - Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (15)

KALINENKO, I.G., kand.biolog.nauk; CHORBA, L.N.

Harvesting time and the quality of strong wheats. Zemledelie 25 no.7:39-41
Jl '63. (MIRA 16:9)
(Caucasus, Northern--Wheat--Harvesting)

KALINENKO, I.G.

A case of proliferation of spikelets in wheat. Bot.zhur. 47
no.3:419-420 Mr '62. (MIRA 15:3)

1. Gosudarstvennaya selektsionnaya stantsiya, g. Zernograd.
(Wheat)

KALINENKO, I.G., kand.biologicheskikh nauk; CHORBA, L.N.; GORYACHEV, N.P.

Strong winter wheats in Rostov Province. Zemledelie 24 no.8:38-41
Ag '62. (MIRA 15:9)

1. Zernogradskaya gosudarstvennaya selektsionnaya stantsiya
Donskogo nauchno-issledovatel'skogo instituta sel'skogo khozyaystva,
(Rostov Province—Wheat—Varieties)

KALINENKO, I.G., kand. biolog. nauk; CHORBA, L.N.

Producing high-grade grain of strong winter wheat and its breeding
for quality. Agrobiologiya 5:774-784 S-0 '64. (MIRA 17:11)

I. Zernogradskaya gosudarstvennaya selektsionnaya stantsiya.

KALINENKO, L.G.; LITVAK, I.M.

Effect of certain factors on the extent of sucrose precipitation
by barium hydroxide. Trudy KTIPP no.22:7-15 '60. (MIRA 14:3)
(Sucrose) (Barium hydroxide)

LITVAK, I.M.; KALINENKO, L.G.

Studying the technological conditions of the settling of the
feed molasse sucrose with barium hydroxide. Trudy KTIPP
no.24:3-7 '61.

(MIRA 15:6)

(Sugar manufacture) (Sucrose)

BRODSKIY, A.M.; KALINENKO, R.A.; LAVROVSKIY, K.P.

Adsorptive analysis and separation of hydrocarbon gases. Khim. i
tekhn. tepl. no.8:18-22 Ag '56. (MIRA 9:10)

1. Institut nefti Akademii nauk SSSR.
(Hydrocarbons)

Kalinenko, R.A.

USSR/Physical Chemistry - Kinetics. Combustion.
Explosives. Topochemistry. Catalysis

B-1

Abs Jour : Referat Zhur - Khimiya, No 2, 1957, 3866

Author : Lavrovskiy K.P., Kalinenko R.A., Rozental' A.L.

Inst : Institute of Petroleum, Academy of Sciences USSR

Title : On Conversion of Hydrocarbon Gases with Steam over
Aluminum-Molybdenum Catalyst.

Craig Pub : Tr. In-ta nefti AN SSSR, 1956, 8, 131-133

Abstract : A study was made of the reaction of conversion of methane, ethane and propane with steam over an aluminum-molybdenum catalyst, in a flow system at 570-620°, space velocities of 12-100 hour⁻¹, and molar ratio H₂O/initial hydrocarbon of 4-19.4. Degree of conversion of CH₄ at a pressure of 1-32 atmospheres does not exceed 2.8%, the degree of conversion of C₂H₆ and of C₃H₈ at a pressure of 1 atmosphere does not exceed, respectively, 14.2 and 27.4%.

Card 1/1

- 153 -

Card 1/2

- 157 -

Shoreline
Marine State Univ.

BRODSKIY, A. I., KALINENKO, R. A., LAVROVSKY, K. P.

"Application of Adsorption Methods to the Analysis and Separation of Gaseous Hydrocarbons in the Study of Kinetics with the Aid of Tracers."

p.399-404,

Problemy Khimii i Katalizma, v. 2, Isotopes in Catalysts, Moscow, Izd-vo AN SSSR, 1957. 140p.

Most of the papers in this collection were presented at the Conference on Isotopes in Catalysts which took place in Moscow, Mar 31 - Apr 5, 1956.

KALINENKO, R.A.

20-5-19/48

AUTHORS: Brodskiy, A. M. , Kalinenko, R. A. , Lavrovskiy, K. P. , Corresponding Member AN USSR, and Titov, V. B.

TITLE: Kinetic Laws in the High-Temperature Cracking of Ethane (O kineticheskikh zakonomernostyakh vysokotemperaturnogo krekinga etana)

PERIODICAL: Doklady AN SSSR, 1957, Vol. 116, Nr 5, pp. 789 - 792 (USSR)

ABSTRACT: In this paper the investigation of the total kinetics of this cracking between 800 and 900° is described. The increase of temperature and the corresponding rapid shortening of the reaction period from 0,5 to 0,005 seconds demand a special experimental method. The experiment was divided into 2 parts: 1.) the cracking itself and 2.) analysis of the products. In the case of the latter a chromatographical method worked out by the authors was used (reference 3), where this method failed because of the small quantity of the single gases (e.g. isobutane), the method of radioactive indicators was used. In addition to that, a small quantity of methane, marked with C¹⁴, was added to the initial ethane. Figure 1 gives the arrangement of the basic elements of the experimental device. During the experiments a "boiling layer" (reference 2) was produced in the reactor. After a quick cooling of the cracking products after the output from the boiling layer CO₂ of room temperature and in

Card 1/3

20-5-19/48

Kinetic Laws in the High-Temperature Cracking of Ethane

measured in the previous paper (reference 6). By means of the authors' method it was found that in the ethane cracking products in tenth % quantities divinyl, butylene, and only traces of isobutane, finally propylene and propane, a fact which was never defined exactly in the references. Figure 2 furthermore shows that the known self-inhibition effect is not expressed up to high degrees of transformation. This can be explained by the connection between the self-inhibition at lower temperature and the influence of the walls. There are 3 figures, 1 table, and 7 references, 4 of which are Slavic.

ASSOCIATION: Petroleum Institute AN USSR
(Institut nefti Akademii nauk SSSR)

SUBMITTED: May 25, 1957

AVAILABLE: Library of Congress

Card 3/3

The Significance of Chain Reactions in the High-Temperature 20-6-26/47
Cracking of Ethane

to the solution of a system of algebraic equations. The author additionally includes 5 elementary processes in the examination. Then the expressions for the dependence of the concentrations of the various active products on time, obtained due to a special analysis, are given. A provisional estimation already shows that the portion of chain reactions in the total process of cracking within the frame of the generally used scheme in the case investigated here is very small. Detailed numerical data on this are given. There are 1 table, and 9 references, 5 of which are Slavic.

ASSOCIATION: Petroleum Institute AN USSR (Institut nefti Akademii nauk SSSR)

SUBMITTED: July 18, 1957

AVAILABLE: Library of Congress

Card 2/2

KALINENKO, R. A., Cand Chem Sci -- (diss) "Study of the mechanism of
ethane cracking at high temperatures ^{by mean^s of tagged atoms} with use of ^{labeled samples.}
Mos, 1958. 10 pp (Acad Sci USSR, Inst of Petroleum), 120 copies (KL,
16-58, 116)

- 14 -

66834

5(4) 5.3200

SOV/76-33-11-13/47

AUTHORS: Brodskiy, A. M., Kalinenko, R. A., Lavrovskiy, K. P.,
Titov, V. B.

TITLE: On the Mechanism of High-temperature Cracking of Ethane

PERIODICAL: Zhurnal fizicheskoy khimii, 1959, Vol 33, Nr 11, pp 2457-2466
(USSR)

ABSTRACT: The reaction mechanism of the cracking of hydrocarbon gases at 770-890° was studied, as in general the industrial pyrolysis of these gases takes place at these high temperatures. A special experimental method was developed by which tracer atoms and ethane are used to which approximately 2% of marked C¹⁴H₄ methane was added. The experiments were made in a continuously working apparatus (Fig 1) at approximately 90 mm Hg. The quartz reactor was filled with corundum acting as heat carrier, and the temperature was recorded by means of an EPP-09 electronic potentiometer. The results obtained (Tables 1-3) showed that at these temperatures the maximum participation of the chain-reaction process in the conversion of ethane into ethylene is 5%, and that the inherent inhibition ✓

Card 1/2

66858

SOV/76-33..11-13/47

On the Mechanism of High-temperature Cracking of Ethane

characteristic of cracking at 500-650°C does not occur. The reaction proceeds according to the first order, and the activation energy is 82 ± 2 kcal/mol. The reaction-rate constant of $\text{CH}_3 + \text{C}_2\text{H}_6 \rightarrow \text{CH}_4 + \text{C}_2\text{H}_5$ points to a steric factor of the order of 10^{-3} for this reaction, while the activation energy obtained from 12 ± 2 kcal is in agreement with data from other publications. The recombination constant, obtained both by experiment and by calculation using thermodynamic data, is approximately $5 \cdot 10^2$ times smaller than that obtained at lower temperatures. The decomposition rate constant of the ethyl radical is 10^3 times smaller than it would be according to the elementary kinetic gas theory. There are 4 figures, 3 tables, and 14 references, 8 of which are Soviet.

ASSOCIATION: Akademiya nauk SSSR, Institut neftekhimicheskogo sinteza
(Academy of Sciences, USSR, Institute of Petroleum-chemical
Synthesis) ✓

Card 2/2

On the Isotope Effect in the Cracking of Ethane

SOV/20-124-2-28/71

(accuracy ~ 1%). The activity of methane is not equal to A/2 (as it would have to be in the case of lacking isotopic effect) but much lower. Here A is the activity of the ethane mixture existing when measurements were begun. A table contains the values of ethane activity in % of A/2 as function of the ethylene content in the cracked gas. Methane activity is lower by ~10% than A/2 and varies relatively little with progressing reaction. The value of the isotopic effect found is near that found previously (Refs 1,2) for propane. The data mentioned above all confirm (on the basis of ethane) the abnormally high value of the isotopic effect in the reaction of methane formation. The equality of the order of magnitude of the isotopic effect (with respect to methane) for C_2H_6 and C_3H_8 indicates the existence of similar ethanes in the formation of CH_4 in the two above-mentioned cases. Correction note: The provisional experiments carried out by the authors concerning the cracking of the ethane mixture $C^{12}H_3 - C^{12}H_3$ and $C^{14}H_3 - C^{14}H_3$ showed that in this case the isotopic effect (with respect to methane) is considerably

Card 2/3

On the Isotope Effect in the Cracking of Ethane

SOV/20-124-2-28/71

lower than in the cracking of $C^{12}H_3 - C^{14}H_3$. This confirms the above assumption that the abnormally high value of the isotopic effect in the cracking of ethane $C^{12}H_3 - C^{14}H_3$ is due to a quantum effect connected with the disturbance of symmetry. There are 1 table and 7 references, 4 of which are Soviet.

SUBMITTED: September 24, 1958

Card 3/3

5(4)

307/20-126-6-41/67

AUTHORS:

Brodskiy, A. M., Kalinenko, R. A., Levrovskiy, K. P.,
Corresponding Member, AS USSR

TITLE:

On the Relation Between the Kinetic Isotopic Effects During
 $C^{12}-C^{14}$ and $C^{14}-C^{14}$ Bond Rupture (O sootnoshenii kineticheskikh
izotopnykh effektov pri razryve svyazey $C^{12}-C^{14}$ i $C^{14}-C^{14}$)

PERIODICAL:

Doklady Akademii nauk SSSR, 1959, Vol 126, Nr 6, pp 1293-1295
(USSR)

ABSTRACT:

The effects mentioned in the title were investigated under
conditions of high temperature cracking by means of a mixture of
 $C^{12}H_3-C^{12}H_3$ with $C^{14}H_3-C^{14}H_3$. The results were compared with

the cracking of $C^{14}H_3-C^{12}H_3$ as described in reference 1.

This experiment was made for the reason that hydrocarbons with
only partly marked C-atoms yielded higher values for the
isotopic effect (Refs 1-3) than could be expected according to
the present theoretical opinions (Refs 4, 5). The following is
given as a possible explanation of this phenomenon:

Card 1/3

On the Relation Between the Kinetic Isotopic Effects SOV/20-126-6-41/67
During C¹²-C¹⁴ and C¹⁴-C¹⁴ Bond Rupture

hydrocarbon molecules exhibit a plane of symmetry vertical to the chain or a corresponding alternating axis. The introduction of a marked C-atom has a disturbing effect upon this symmetry, and the transition from the symmetrical to the asymmetrical molecule might entail a considerable kinetic effect. Herefrom it resulted that the symmetrical ethanes C¹²H₃-C¹²H₃ and C¹⁴H₃-C¹⁴H₃ had to differ from asymmetrical

C¹²H₃-C¹⁴H₃ in their effect. The experimental data (Table 1) shows that the kinetic isotopic effect amounts to 5±1% in the formation of methane from C¹⁴H₃-C¹⁴H₃; it is, therefore, considerably lower than the value of 12±2% of reference 1 found for asymmetrical ethane. Measurements were made under entirely equal conditions. This result shows that there is no direct proportion between the kinetic isotopic effect and the reduced mass, and confirms the assumption that the disturbance of the

Card 2/3

5(4)

AUTHORS:

Brodskiy, A. M., Kalinenko, R. A.,
Lavrovskiy, K. P.

S/076/60/034/01/031/044

B004/B007

TITLE:

A Method of Investigating the Mechanism of Fast Reactions in
a Turbulent Reactor by Means of Tagged Atoms ✓

PERIODICAL:

Zhurnal fizicheskoy khimii, 1960, Vol 34, Nr 1, pp 192 - 195
(USSR)

ABSTRACT:

The term turbulent reactor is applied by the authors to a reactor vessel with an intense intermixing device. After giving a survey of the methods of investigating the kinetics of reactions and mentioning the isotopic method by M.B.Neyman, the authors explain the simplification of kinetic equations attained when using tagged atoms and by intensively intermixing the reagents as a result of the falling away of concentration- and temperature gradients. The method makes it possible to deal with both single processes in the reaction and also with the total process, and may therefore be used for the purpose of judging the correctness of the scheme upon which the reaction is based. By the example of the cracking of

Card 1/2

A Method of Investigating the Mechanism of Fast Reactions in a Turbulent Reactor by Means of Tagged Atoms

S/076/60/034/01/031/044

B004/B007

C_2H_6 at 800 - 900° it is shown that by thoroughly intermixing, linear dependence was obtained for the function $y = f(t)$ (Figs 1, 2). There are 2 figures and 6 Soviet references.



SUBMITTED: April 10, 1959

Card 2/2

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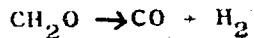
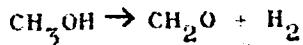
31091
S/195/61/002/004/007/008
E030/E585

AUTHORS: Brodskiy, A. M., Kalinenko, R. A., Lavrovskiy, K. P.,
and Shevel'kova, L. V.

TITLE: Principles of the decomposition of methanol at high
temperatures

PERIODICAL: Kinetika i kataliz, v.2, no.4, 1961, 553-561

TEXT: Previous investigations of the decomposition of
alcohols from C₂ to C₄ postulated an approximately first-order
reaction, involving rupture of C-C or C-H bonds but the yields
and mass balances of C, H, and O have disagreed by about 50% and
the activation energy for reaction velocity has been many times
smaller than that for pressure decrease in the system. Decomposi-
tion of methanol was considered by C. J. M. Fletcher (Ref. 6
Proc. Roy. Soc., A147 119, 1954) to be two-stage:



with similar discrepancies. The present work studied the reaction
Card 1/4

31091

Principles of the decomposition

S/195/61/002/004/007/008
E030/E985

gases twofold. In all cases the reaction products had significant concentration of C_2H_6 , CH_3OCH_3 , $C_2H_5OCH_3$, $C_2H_5OC_2H_5$, CH_3CHO , CH_3COCH_3 etc., signifying extensive free radical formation. Moreover, thermodynamic data on the decomposition of methanol predict reaction velocities some two or three orders of magnitude less than observed, so one must be dealing in practice with the formation of free radicals by a highly developed chain reaction. To support this, high concentrations of ethylene were found (20-50% of ethane) and it is known that in the 634-734°C region there is insignificant cracking of methane; the only alternative plausible source is from recombination of CH_3 radicals. V.V. Voyevodskiy is mentioned in the article for his contribution in this field. Acknowledgments are expressed to N.N. Naymushin for his assistance. There are 3 figures, 6 tables and 16 references. 5 Soviet-bloc and 11 non-Soviet-bloc. The four latest English-language references read as follows: Ref. 1: J.A.Barnard, H.W.D. Hughes, Trans. Faraday Soc., 56, 55, 1960; Ref. 2: Ibid, 56, 64, 1960; Ref. 3: J.A.Barnard, Thid, 56, 72, 1960; Ref. 5:Ibid, 55, 947, 1959.

Card 3/4

S/020/62/144/004/018/024
B101/B13B

3309
AUTHORS: Brodskiy, A. M., Kalinenko, R. A., Lavrovskiy, K. P.,
Corresponding Member AS USSR, and Shevel'kova, L. V.

TITLE: Mechanism of by-product formation in high-temperature
cracking of ethane

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 144, no. 4, 1962, 817-820

TEXT: Following previous papers and using techniques described therein
(ZhFKh, 33, no. 11 (1959); ibid., 34, no. 1 (1960)) the formation of
 CH_4 , C_2H_2 , C_3H_8 , C_3H_6 , C_4H_{10} , C_4H_8 , and C_4H_6 during the cracking of ethane
at 800-880°C and 90 ± 3 mm Hg with additional 0.45% of ethylene tagged by C^{14}
was examined. Corundum or ground quartz was used as a heat carrier. The
reaction products were separated by chromatography and their radioactivity
was measured. Results: (1) CH_4 showed low activity, indicating that it
is formed mainly from C_2H_4 of low activity and from transformation products
thereof. About one-half of the CH_4 is formed without the participation of

Card 1/3

Mechanism of by-product formation ...

S/020/62/144/004/018/024
B101/B138

CH_3^* by the decay of high-molecular products. (2) The equal degree of activity exhibited by C_2H_2 and C_2H_4 indicates that C_2H_2 is formed with the participation of a C_2H_4 molecule. (3) C_3H_8 and C_4H_{10} had a low content of C^{14} . They are formed by recombination of weakly active CH_3 and C_2H_5 radicals. (4) C_3H_6 and C_4H_8 showed the same activity as C_2H_4 . They are not formed from C_3H_8 and C_4H_{10} , respectively, but mainly by the disintegration of C_4H_9 and, at temperatures $< 880^\circ\text{C}$, also by C_2H_3 recombining with CH_3 or C_2H_5 . (5) The fact that C_4H_6 (divinyl) is twice as active as C_2H_4 justifies the supposition that it is formed with the participation of 2 molecules of C_2H_4 . As $[\text{C}_4\text{H}_6]$ is larger than corresponds to the equilibrium concentration in the reaction $\text{C}_4\text{H}_6 \rightleftharpoons \text{C}_2\text{H}_2 + \text{C}_2\text{H}_4$, a complex reaction involving free radicals is assumed. (6) The specific activity of the coke at 880°C amounted to one-half the activity of C_2H_4 . At this

Card 2/3

L-36482-65 EPF(c)/EPR/EPT(j)/EWT(m) Ps-4/Pr-4/Ps-4 R21 1/1/97

ACCESSION NR: AP5010561

UR/0204/64/004/005/0691/0699

AUTHOR: Yampolskiy, V. I.; Brodskiy, A. M.; Kalinovskiy, R. A.; Lavyrovskiy, K. P.

TITLE: Transformation of ethylene at high temperatures

SOURCE: Neftekhimika, v. 4, no. 5, 1964, 591-599

TOPIC TAGS: ethylene, high temperature phenomenon, reaction mechanism, chemical kinetics

ABSTRACT: The kinetics and mechanism of the thermal transformations of ethylene were studied in a flow reactor within the temperature range 300°-1000°K. The

Card 1/2

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ACCESSION NR: AP5010561

carrying out the experiments, and to N. Ya. Charnyak for the assistance in the identification of vinyl-acetylene and cyclopentadiene by the method of mass-spectrometry. Order sent from 1 Glimra 8 'Ottobre' 5 yearns 4 mahlac.

"APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000620030005-5

SUBMITTED: 12Mar64

ENCL: 00

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NO REF SOV: 005

OTHER: 012

JPRS

Card 2/2

APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000620030005-5"

SHEVEL'KOV, L.V.; BROPISKY, A.M.; KALINENKO, R.A.; LAUROVSKII, K.P.

Mechanism of the formation of some secondary products in the high-
temperature cracking of ethane. Kin. i kat. 6 no.4:592-600 JI-Ag
'65. (MIRA 18:9)

1. Institut neftekhimicheskogo sinteza imeni A.V.Topchiyeva AN
SSSR.

KALINENKO, R.A.; BRODSKIY, A.M.

Kinetic scheme of ethane pyrolysis for optimization of
ethylene production processes. Kin.i kat. 6 no.5:916-921
S-0 '65. (MIR 18:11)

1. Institut neftekhimicheskogo sinteza imeni Topchiyeva
AN SSSR.